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## Phosphorus, Sulfur, and Silicon and the Related Elements

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### Oxidative Addition of Phospholes and Phosphabenzene into Triosmium and Triruthenium Clusters. Ring Opening Versus C-H Bond Cleavage

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OXIDATIVE ADDITION OF PHOSPHOLES AND PHOSPHABENZENE  
INTO TRIOSMIUM AND TRIRUTHENIUM CLUSTERS.  
RING OPENING VERSUS C-H BOND CLEAVAGE.

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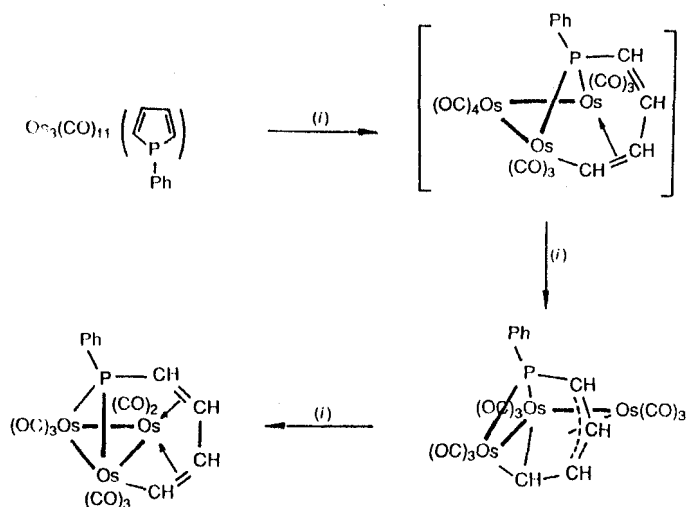
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**Abstract** The five-membered heterocycles compounds 3,4-dimethyl-1-phenylphosphole ( $\text{PhPCH}=\text{CMeCMe}=\text{CH}$ ) and 1-phenylphosphole ( $\text{PhPCH}=\text{CHCH}=\text{CH}$ ) react with the clusters  $[\text{Os}_3(\text{CO})_{11}(\text{MeCN})]$  or  $[\text{Os}_3(\text{CO})_{10}(\text{MeCN})_2]$  to give the simple substitution products  $[\text{Os}_3(\text{CO})_{12-x}(\text{phosphole})_x]$  ( $x=1$  or  $2$ ), in which the phospholes are coordinated as tertiary phosphines through the phosphorus atoms. The six membered heterocyclic compounds phosphabenzene and 2-terbutyl-phosphabenzene react with  $[\text{Os}_3(\text{CO})_{11}(\text{MeCN})]$  or  $[\text{Os}_3(\text{CO})_{10}(\text{MeCN})_2]$  to give simple substitution products  $[\text{Os}_3(\text{CO})_{12-x}(\text{phosphabenzene})_x]$  ( $x=1$  or  $2$ ).  $[\text{Ru}_3(\text{CO})_{12}]$  and  $[\text{Os}_3\text{H}_2(\text{CO})_{10}]$  react with 2-terbutyl-phosphabenzene to give the nonacarbonyl species  $[\text{HRu}_3(\text{CO})_9(\text{P}=\text{CCH}=\text{CH}-\text{CH}=\text{C}(\text{tBu}))]$  and  $[\text{H}_2\text{Os}_3(\text{CO})_9(\text{PCHCH}=\text{CHCH}=\text{C}(\text{tBu}))]$  respectively, where a C-H bond cleavage has occurred and the phosphorus atom is coordinated to a metal center, even though the clusters show different geometries, which are discussed.

1-Phenylphosphole ( $\text{PhPC}_4\text{H}_4$ ) reacts with  $[\text{Os}_3(\text{CO})_{11}(\text{MeCN})]$  to give  $[\text{Os}_3(\text{CO})_{11}(\text{PhPC}_4\text{H}_4)]$  **1** which thermally decarbonylates to the nonacarbonyl  $[\text{Os}_3(\text{CO})_9(\text{PhPC}_4\text{H}_4)]$  **2** and the octacarbonyl  $[\text{Os}_3(\text{CO})_8(\text{PhPC}_4\text{H}_4)]$  **3**, which are shown spectroscopically and by X-ray diffraction methods to contain ring-opened ligands. A decacarbonyl intermediate was not observed but is believed to have the structure shown in Scheme 1 on the basis of the known structure of the related compound

formed by ring-opening of selenophene<sup>1</sup>. In each decarbonylated product which we isolated, the  $\text{PhPC}_4\text{H}_4$  ligand behaves as an 8-electron donor, but the modes of bonding in the two compounds are very different<sup>2</sup>.



SCHEME 1.

Both acetonitrile complexes  $[\text{Os}_3(\text{CO})_{12-x}(\text{MeCN})_x]$  ( $x = 1$  or  $2$ ) react with 3,4-dimethyl-1-phenylphosphole ( $\text{PhPC}_4\text{H}_2\text{Me}_2$ ) to give in different amounts the two complexes  $[\text{Os}_3(\text{CO})_{12-x}(\text{PhPC}_4\text{H}_2\text{Me}_2)_x]$  **4** and **5** ( $x = 1$  or  $2$  respectively). Thermal decarbonylation of  $[\text{Os}_3(\text{CO})_{11}(\text{PhPC}_4\text{H}_2\text{Me}_2)]$  **4**, gave purple crystals of the nonacarbonyl  $[\text{Os}_3(\text{CO})_9(\text{PhPC}_4\text{H}_2\text{Me}_2)]$  **A**, (44% conversion) and two other yellow compounds in less than 2% conversion. One of these was characterised mainly by X-ray diffraction as  $[\text{Os}_3(\mu\text{-H})(\mu\text{-PC}_4\text{H}_2\text{Me}_2)(\mu\text{-C}_6\text{H}_4)(\text{CO})_9]$  **B**, an isomer of **A**. The other minor yellow product **C** is also an isomer of **A**, but unlike **B**, it is converted readily into **A** on standing. Because of this, **C** was only characterised spectroscopically. We have found that the purple isomer **A** is converted into this other yellow isomer **C** on irradiation with white light from a fluorescent tube. Isomer **C** is converted into a mixture containing predominantly isomer **A** over several hours at room temperature in the dark or under low intensity light conditions. The cluster  $[\text{Os}_3(\text{CO})_9(\text{PhPC}_4\text{H}_2\text{Me}_2)]$  **A** is not a hydride derivative ( $^1\text{H}$  nmr evidence) and as well as phenyl resonances in the

$^1\text{H}$  nmr spectrum, four other signals were observed (Me signals at  $\delta$  1.07 and 2.30; C-H signals at  $\delta$  2.82 and 11.70). Decarbonylation has led to major changes in the bonding of the phosphole ligand, but the product does not correspond spectroscopically at all to the unsubstituted phosphole derivative  $[\text{Os}_3(\text{CO})_9(\text{PhPC}_4\text{H}_4)]$  **2**. The molecular structure of **A** is shown in Fig. 1.

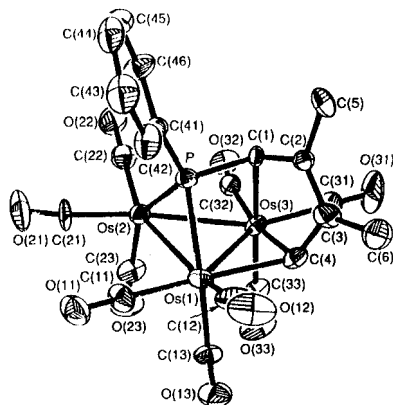


FIGURE 1. Molecular Structure of  $[\text{Os}_3(\text{CO})_9(\text{PhPC}_4\text{H}_2\text{Me}_2)]$ , cluster **A**.

The cluster  $[\text{H}_2\text{Os}_3(\text{CO})_{10}]$  reacts with phosphabenzene and 2-*tert*-butyl-phosphabenzene ( $\text{RC}_5\text{H}_4\text{P}$ ,  $\text{R}=\text{H}$  or  $t\text{Bu}$ ) in refluxing cyclohexane to give the clusters containing a new type of triply bridging phosphabenzene ligand,  $[\text{Os}_3(\mu\text{-H})_2(\mu_3\text{-RC}_5\text{H}_4\text{P})(\text{CO})_9]$  **6**, and the phosphabenzene coupled product  $[\text{Os}_3(\mu\text{-H})(\mu_3\text{-C}_{10}\text{H}_{11}\text{P}_2)(\text{CO})_9]$  **7**. (Figure 2).

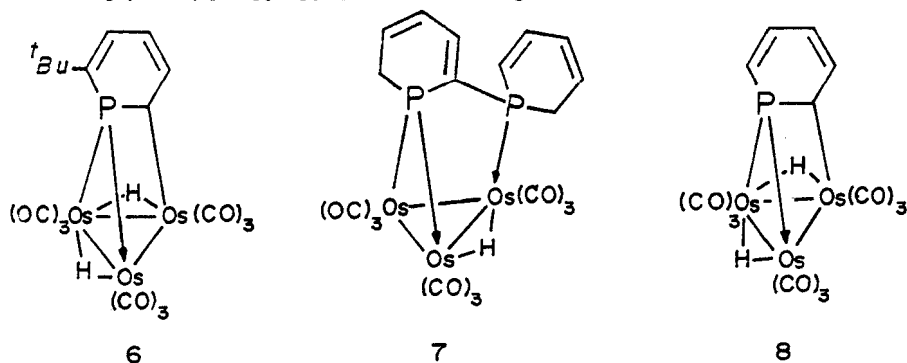


FIGURE 2. Structures of the phosphabenzene-derived triosmium clusters **6**, **7** and **8**.

Clusters **6** and **8** have almost identical IR data around  $2000\text{ cm}^{-1}$  and hence, directly related structures. The single X-ray crystal structure of cluster **8** has been determined, which shows the phosphabenzene as a  $\mu^3$ -, 4-electron donor, and the coordinatively saturated phosphorus atom behaves as a 3-electron donating phosphido bridge between two osmium atoms; the *alpha* carbon is *sigma* bonded to the third osmium atom. This leaves the rest of the ring as a diene unit, and the ring is distinctly non-planar as a result of the tetrahedral geometries at the phosphorus and at the *alpha* carbon.

Cluster **7** is derived by coupling of two phosphabenzene ligands, giving a new diphosphine ligand which is bonded to the metal atoms through a phosphido-bridging atom and through a tertiary phosphine atom, and behaves as a 5-electron donor. The mechanism of formation of the diphosphine ligand is unclear and involves various H-atom shifts as well as the formation of the P-C bond.

$[\text{Ru}_3(\text{CO})_{12}]$  reacts smoothly with 2-terbutyl-phosphabenzene to give in reasonable yields  $[\text{Ru}_3(\mu\text{-H})(\mu_3\text{-PC}_5\text{H}_4^t\text{Bu})](\text{CO})_9$  **9**, in which C-H bond cleavage occurs. The ligand in this case acts as a five-electron donor, (Fig. 3).

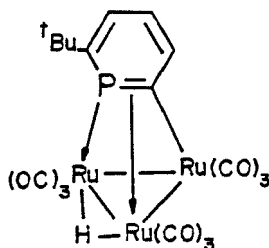


FIGURE 3. Structure of  $[\text{Ru}_3(\mu\text{-H})(\mu_3\text{-PC}_5\text{H}_4^t\text{Bu})(\text{CO})_9]$ , cluster **9**.

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